

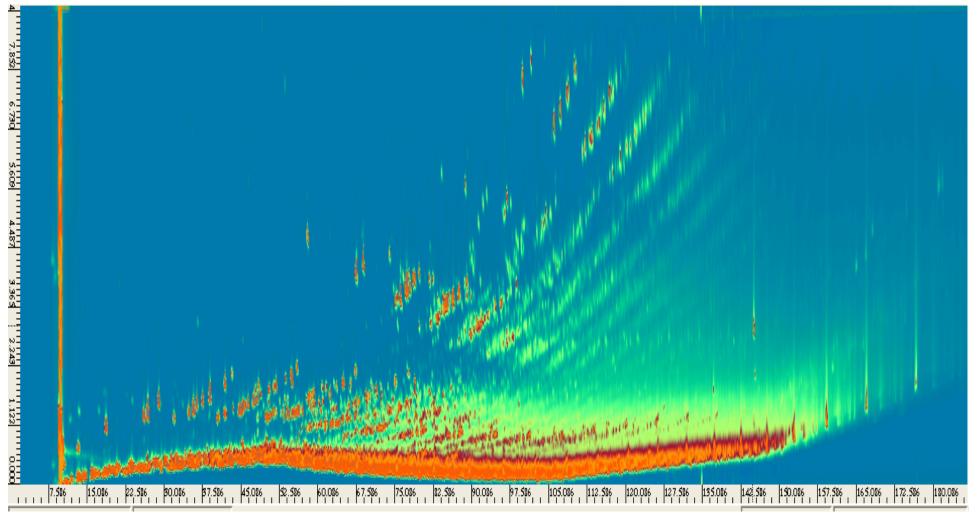
Characterization of Hydrocarbons in petroleum fractions using Comprehensive Gas-Chromatography GCxGC coupled with Fast Scan Quadrupole MSD

Marc Gibert - Ingenieria Anal´ñitica s.L.





Why GCxGC-MSD in petrochemistry







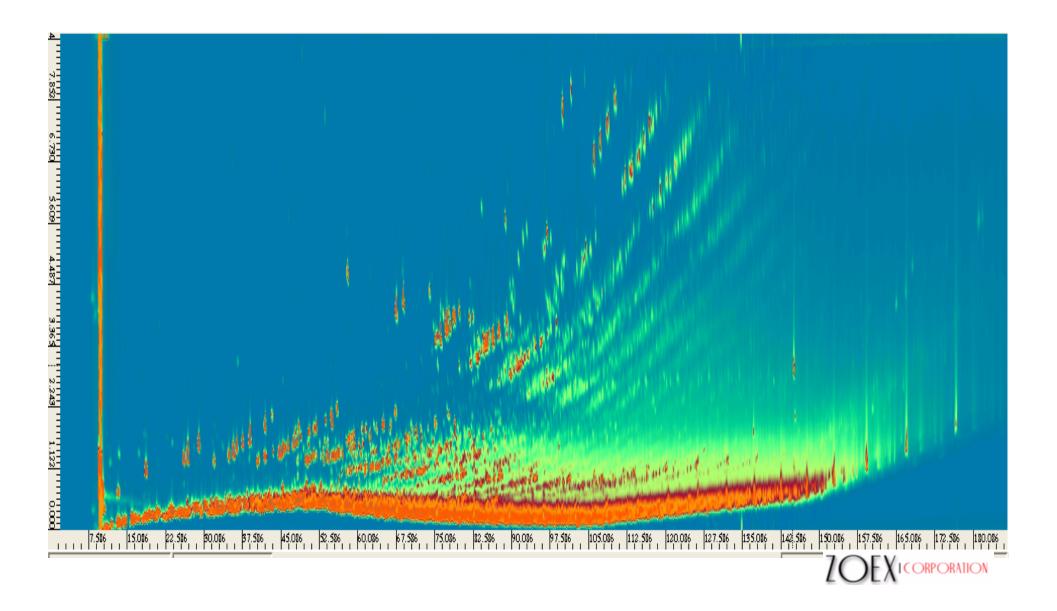
Which information is in this picture

 Fingerprint of crude oils (based on bidimensional distribution of more than 1000 compounds)



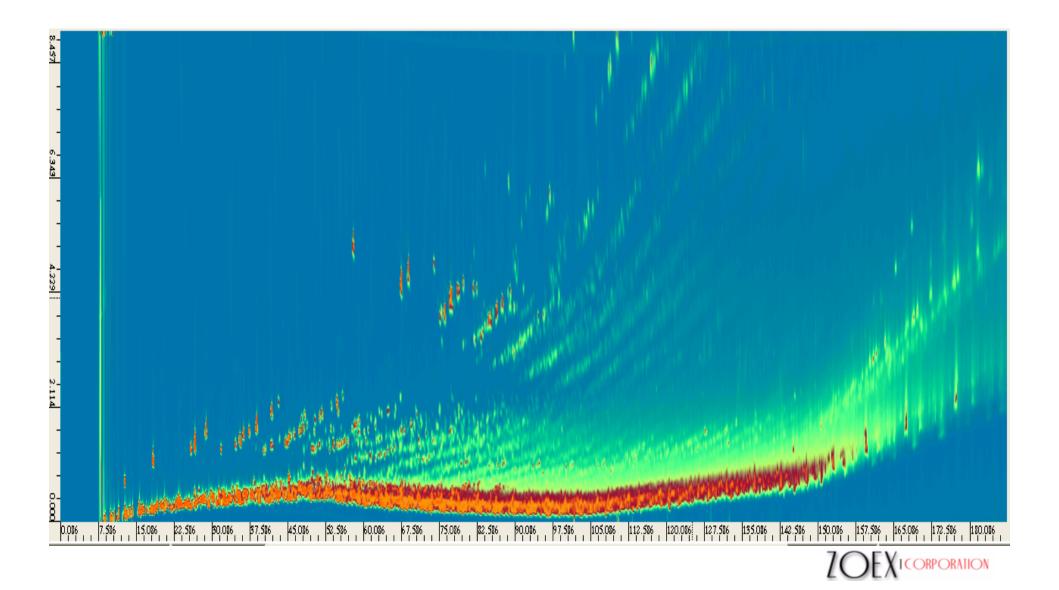


Light crude oil A



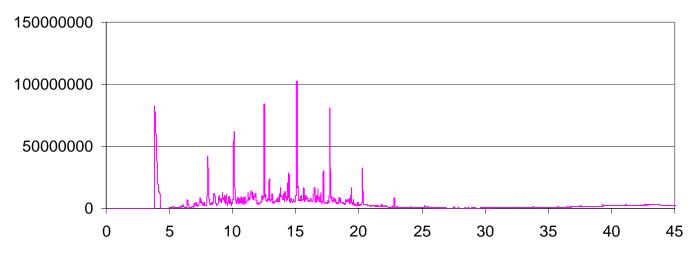


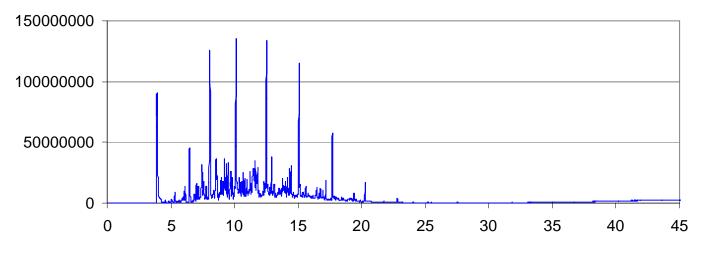
Light crude oil C





Light crude oil GC-MS chromatograms comparison









Which information is in this picture

- Fingerprint of crude oils (based on bidimensional distribution of more than 1000 compounds)
- Chemical information (chemical structure, volatility, polarity, etc.)





Volatility vs. Polarity

ANr name	RT1; RT2 (min)	b.p.°C	polarity index
1 n-hexane	10.436; 0.715	69.0	0.1
2 benzene	12.536; 1.277	80.1	2.7
3 cyclohexane	12.536; 0.766	80.7	0.2
4 n-heptane	14.036; 0.817	98.4	0.0
5 cyclohexane, meth	yl 15.386; 0.868	101.0	n.d.
6 toluene	18.086; 1.635	111.0	2.4
Polarity Delarity	6 4 5 0 Volatility	4	
7.586 9.086 10.586 12.086 13.586 15.086 16.586 18.086 19.586 21.086 22.586 24.086 25.586 ZOEX CORPORATION			



Which information is in this picture

- Fingerprint of crude oils (based on bidimensional distribution of more than 1000 compounds)
- Chemical information (chemical structure, volatility, polarity, etc.)
- Information for application purposes (compound identification for geological exploration and process refinery uses)





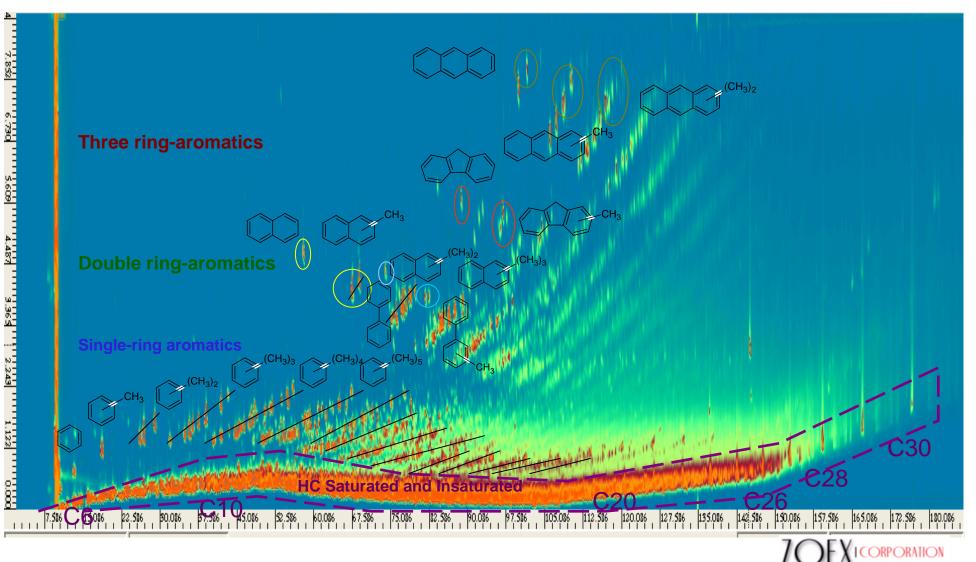
GC x GC method approach

- <u>Goal</u>: use the potentiality of the GCxGC to obtain as much as possible information and semplify instrument use for routine analysis
- <u>One GCxGC method is here described:</u>
 - to optimise the separation among classes;
 - to discrimine among information the more interesting ones:
 - -Hydrocarbons group type
 - -S-containing compounds
 - -N-containing compounds



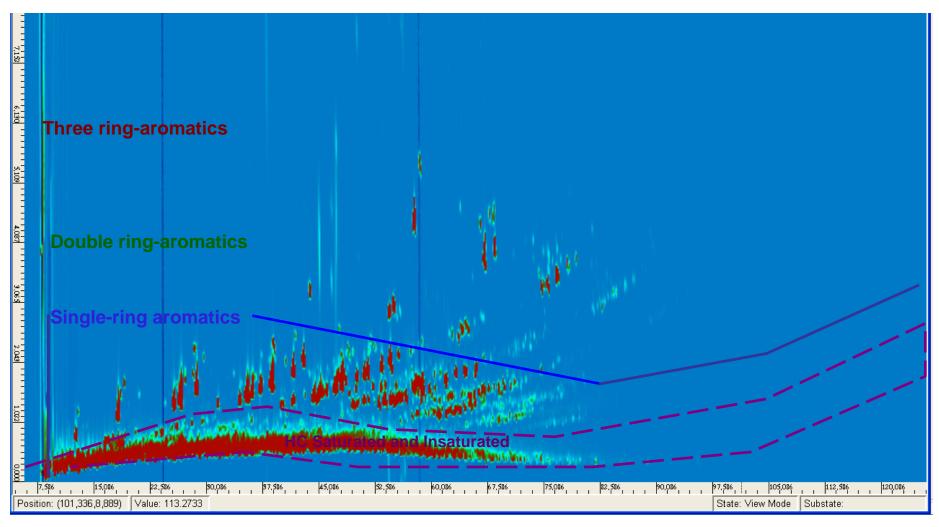


Hydrocarbons in light crude oil A





GCxGC analysis of light petroleum distillate





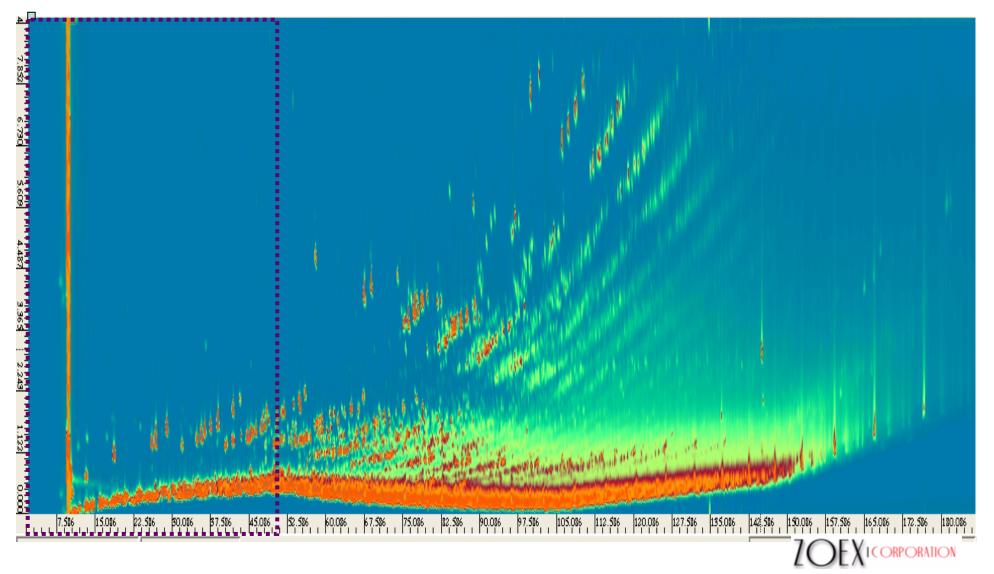
Why to study sulphur compounds

- The increasing concern on the environmental conditions requires to limit the sulphur level in diesel fuel and gasoline (e.g. the specification in Europe for S content is lower than 10 ppm S).
- Very deep conversion of sulphur compounds is therefore required to produce extremely clean transportation fuel. Nature and amount of the sulphur compounds strongly influence the conditions of desulphurisation treatments.
- The understanding and mapping of these species is of vital importance for the further treatment of oil and oil derivatives.



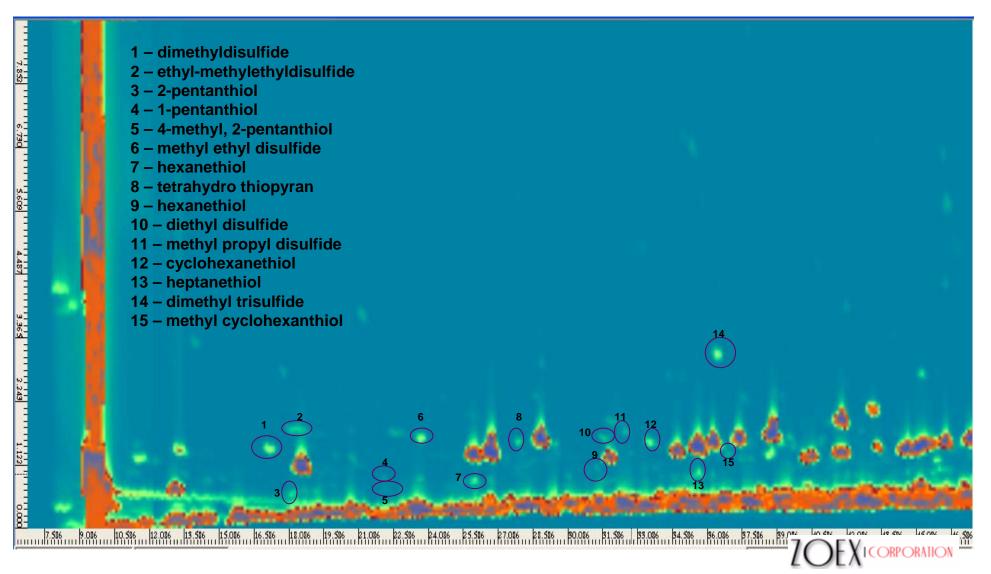


S-containing compounds in light crude oil A



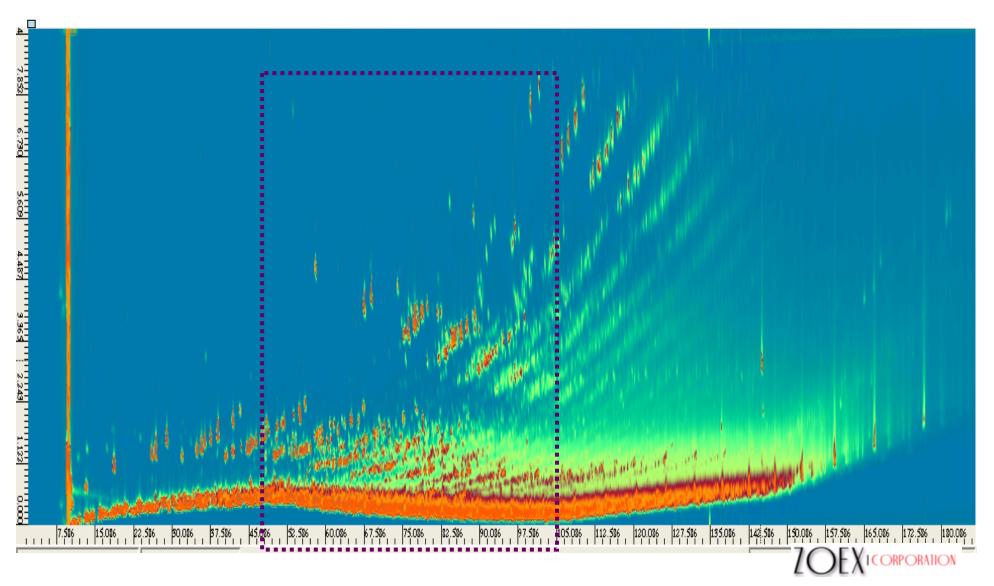


Thioles and sulfides in light crude oil A



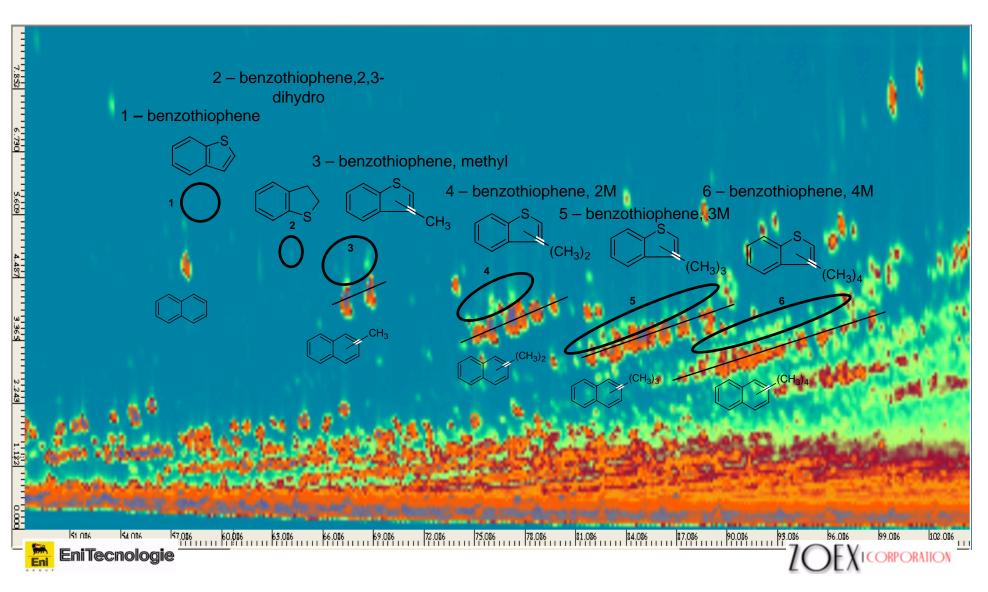


S-compounds in light crude oil A



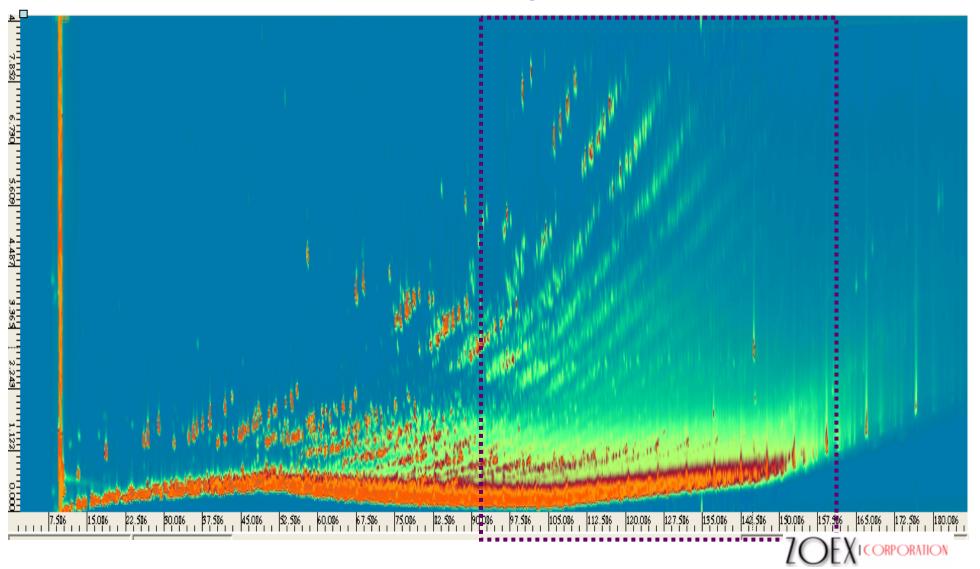


Benzothiophenes in light crude oil A



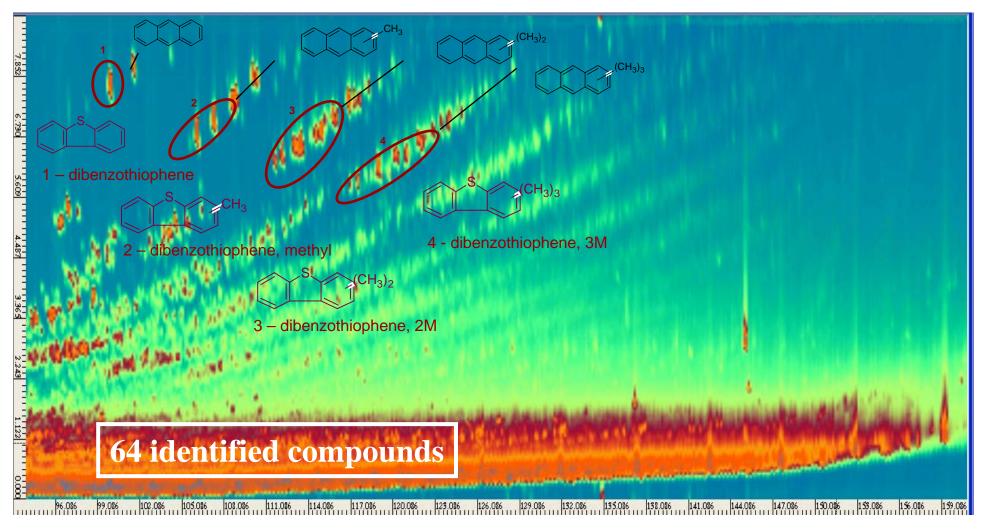


S-compounds in light crude oil A





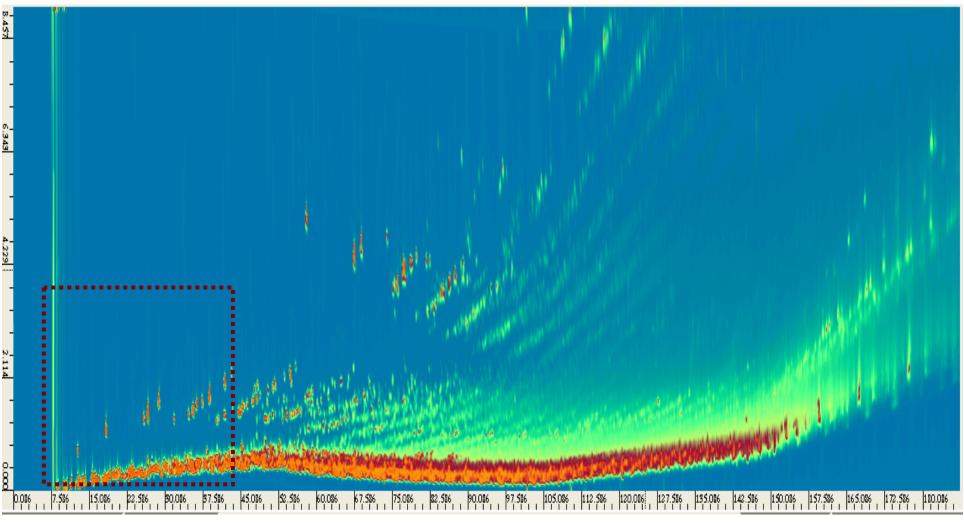
Dibenzothiophenes in light crude oil A







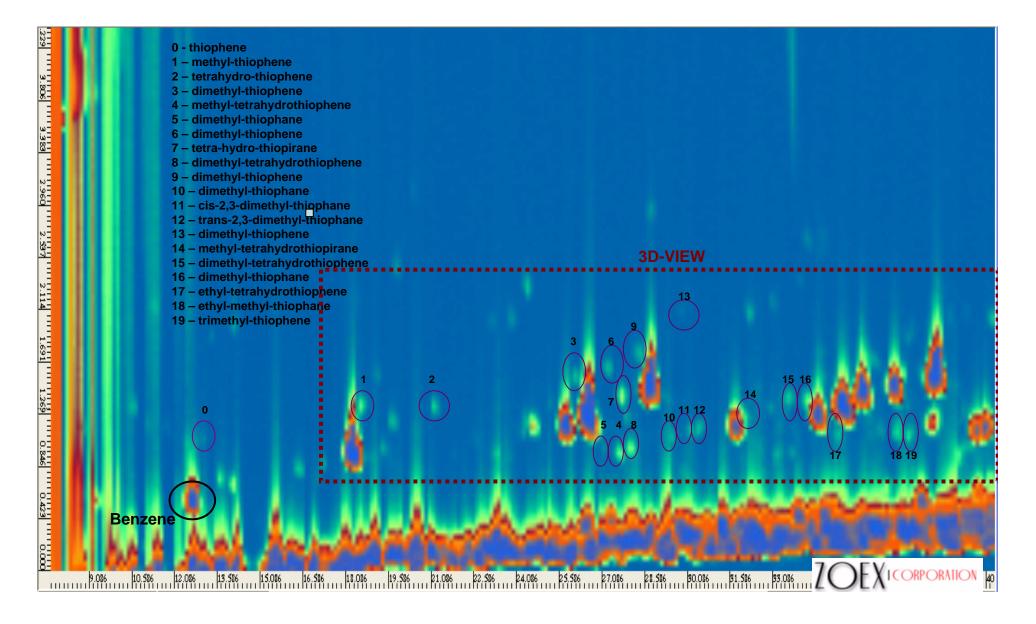
S-compounds in light crude oil C





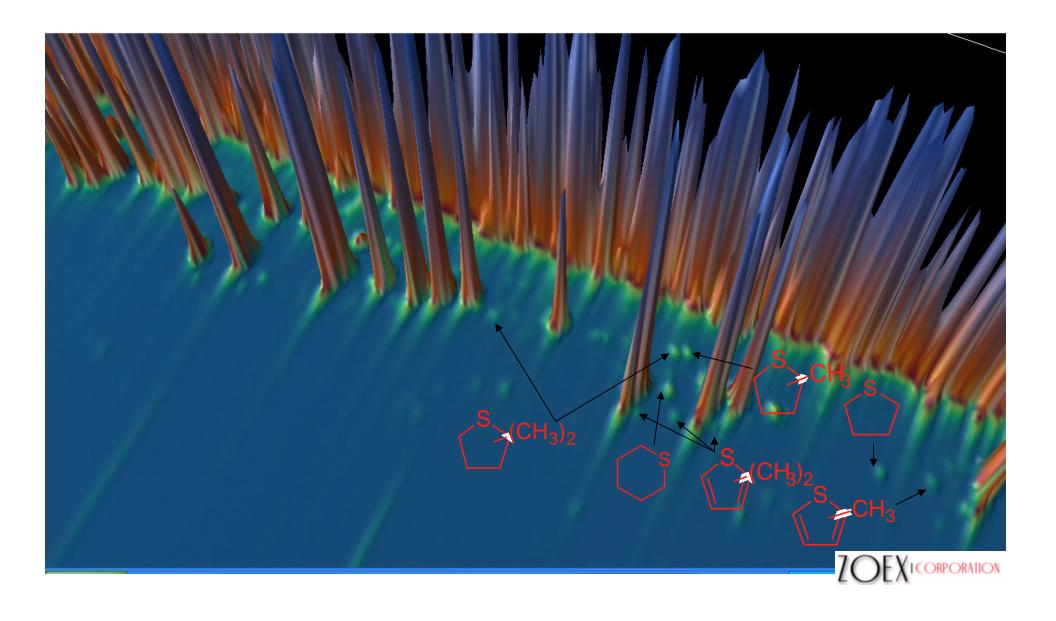


Thiophenes in light crude oil C



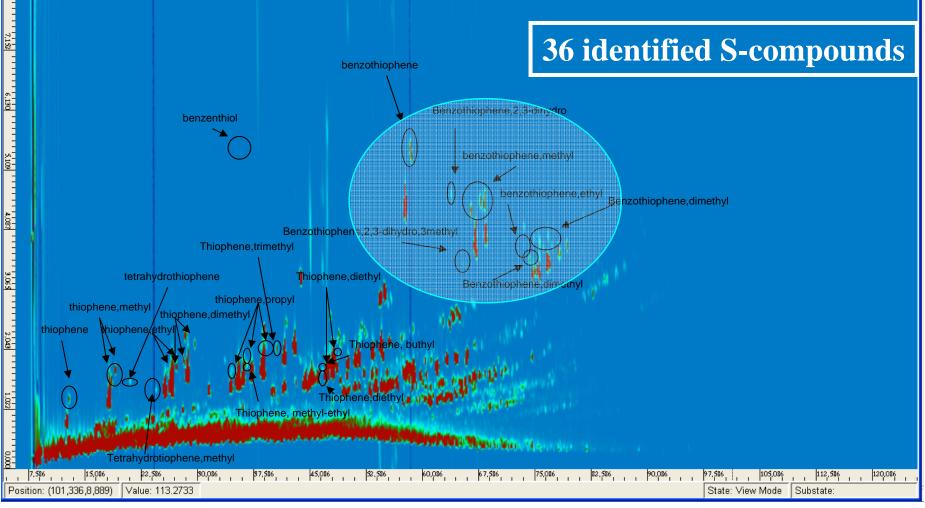


3D-view: Thiophenes in light crude oil C





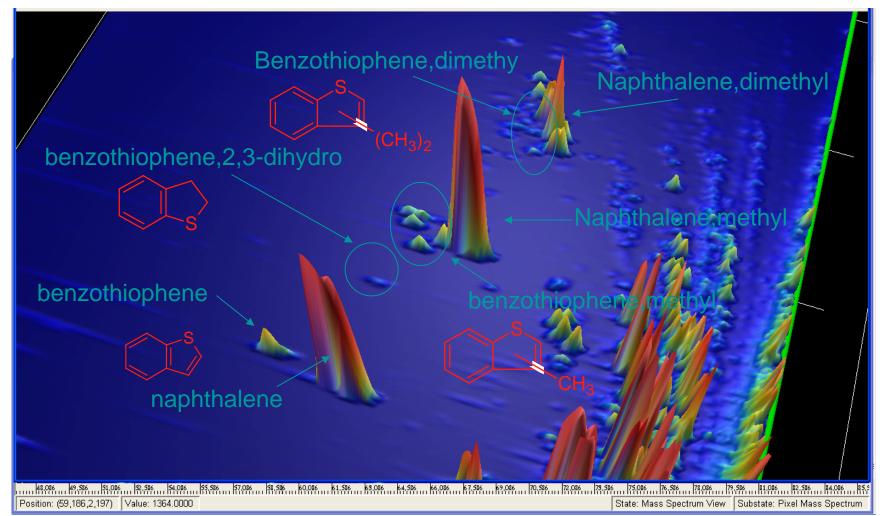
GCxGC S-compound map of light petroleum distillate (527 ppm S)







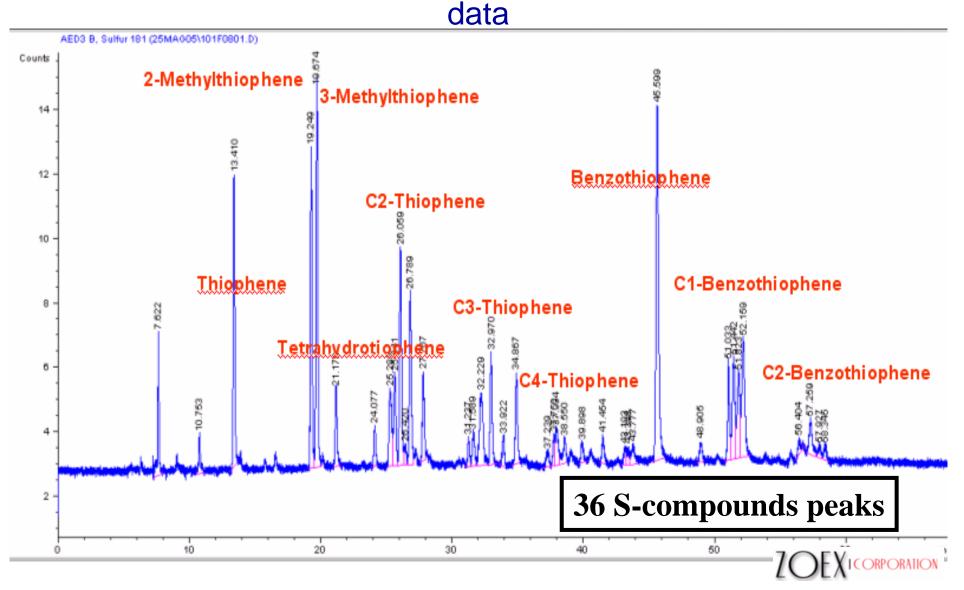
Benzothiophenes 3D-view





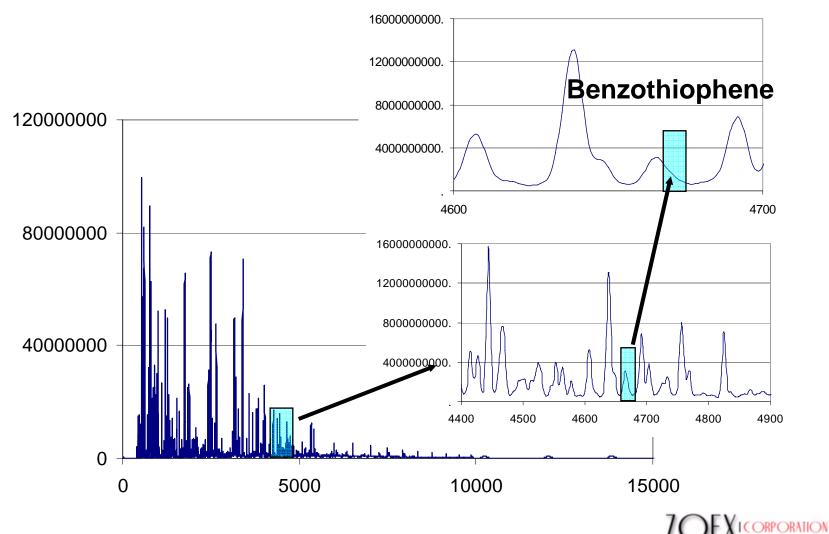


S-compound identification: comparison with GC-AED



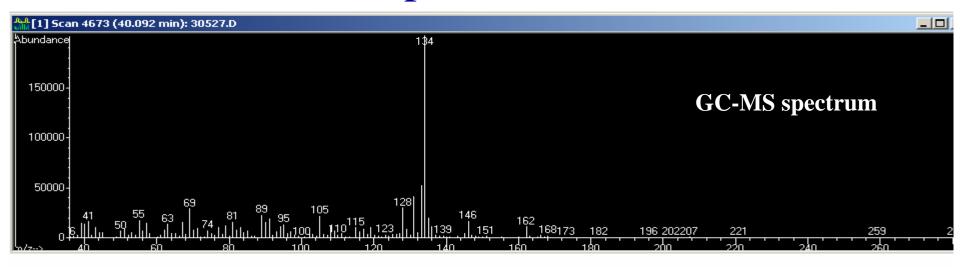


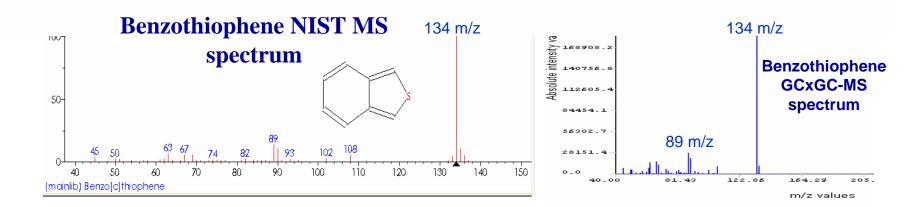
GC-MS analysis





Benzothiophene GC-MS









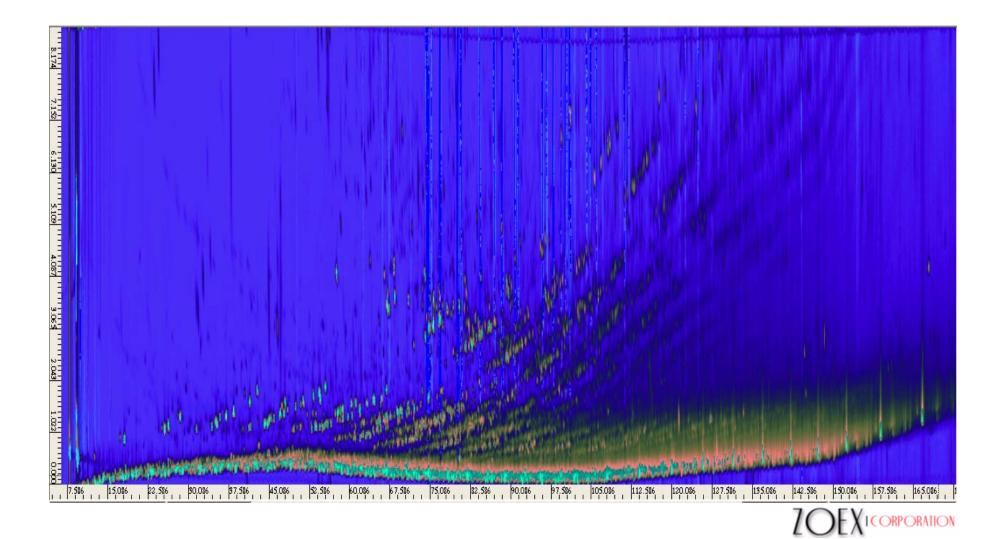
Why to study nitrogen compounds

- During desulphurisation, the presence of refractory and competitive species as nitrogen compounds is highly detrimental, due to their effect as inhibitors for HDS (hydrodesulphurisation) catalysts.
- Neutral (e.g. carbazole), slightly basic (e.g. quinoline) and basic (e.g. pyridine) nitrogen compounds are present in gas oil and cracked oil stream. The need to pre-remove these species is a function of their nature, the more stringent the removal the more basic are the species.
- The mapping of the nitrogen species is of vital importance for the further treatment of oil and oil derivatives.





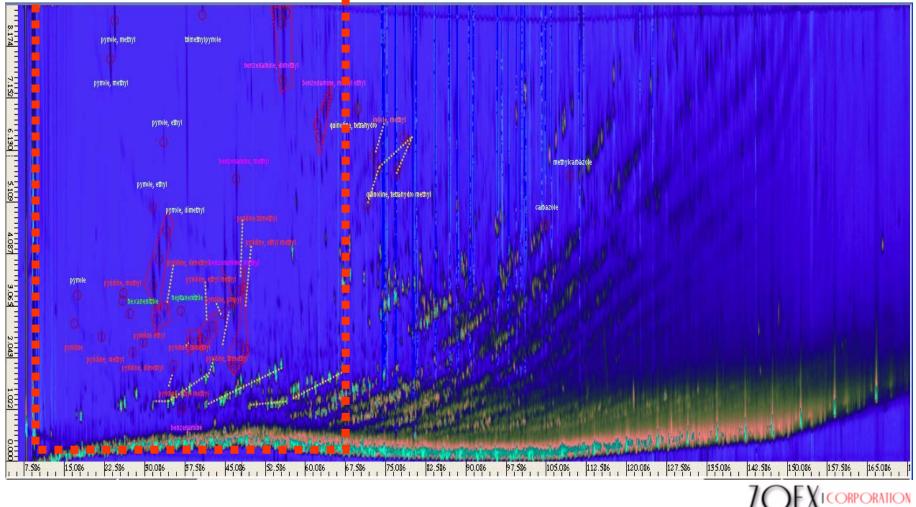
N-compounds in Naphtha





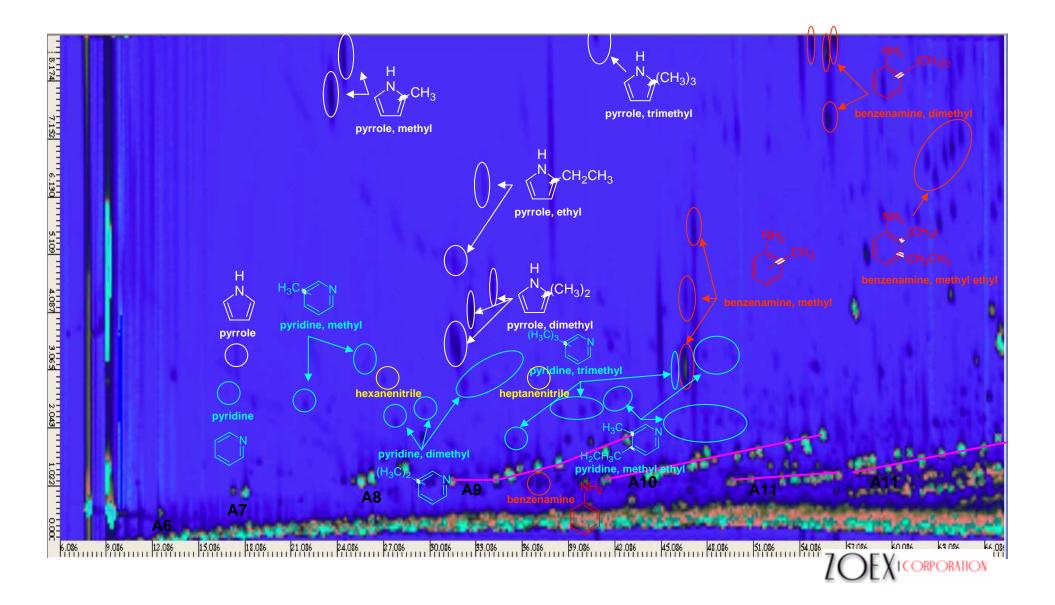
N-compounds in Naphtha





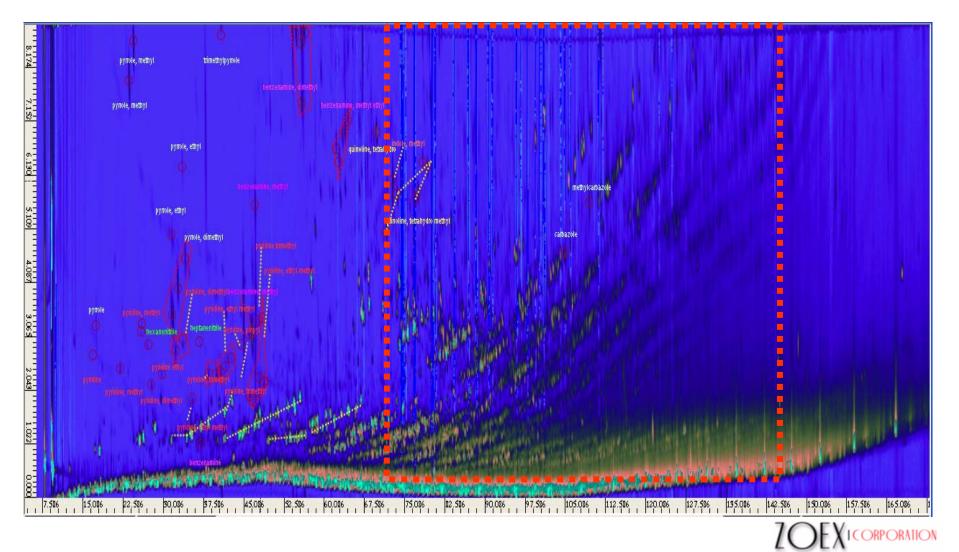


N-compounds in Petroleum fraction



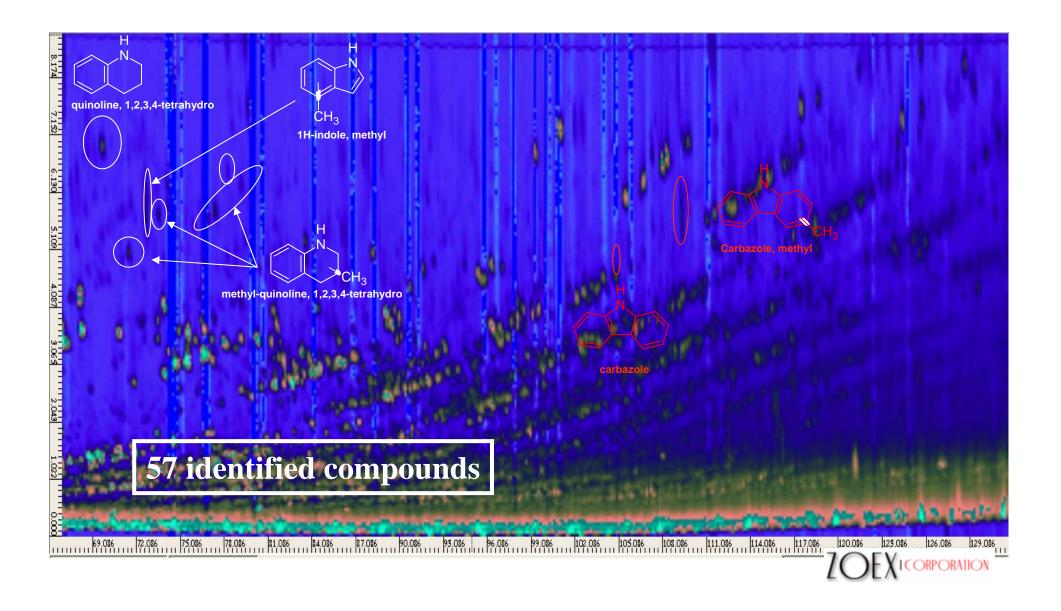


N-compounds in Petroleum fraction



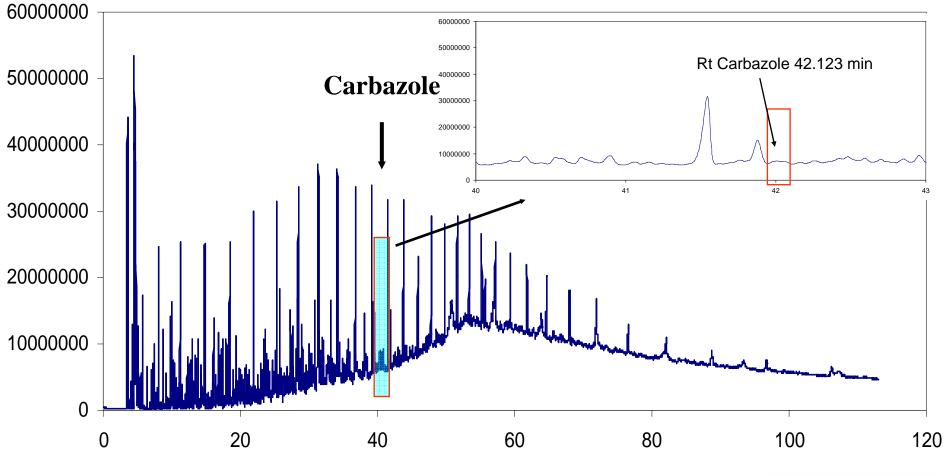


N-compounds in Petroleum fraction



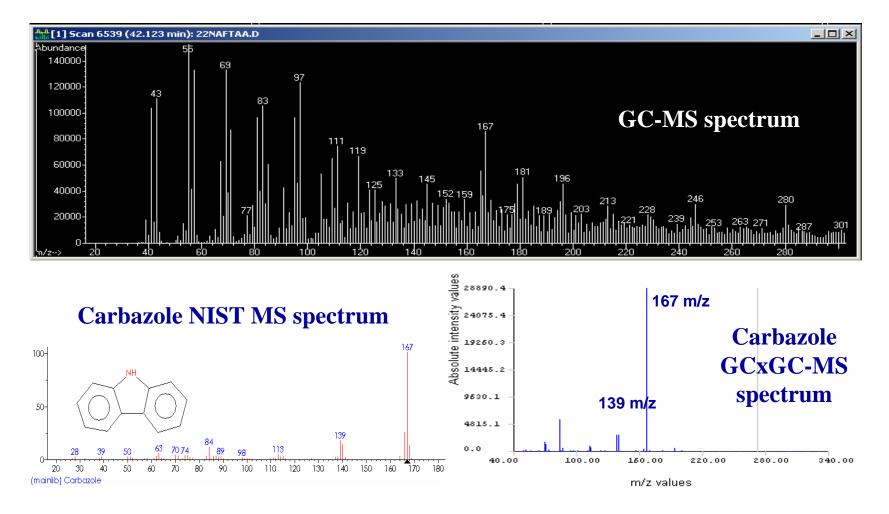


GC-MS analysis





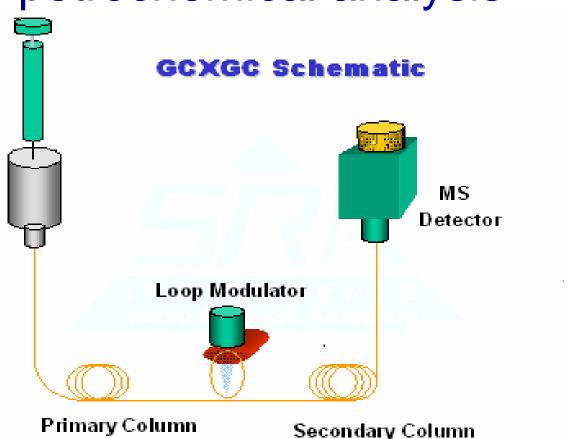
Carbazole MS Spectra







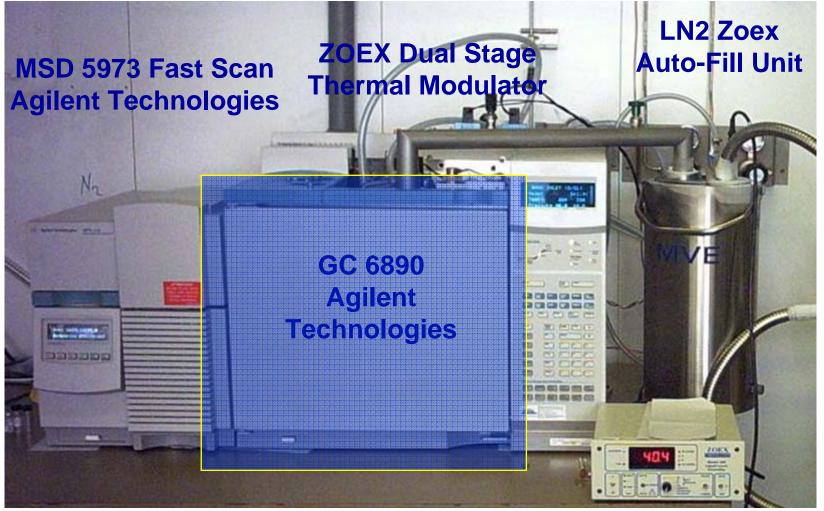
Development of one GCxGC method for petrochemical analysis







GCxGC-MSD System







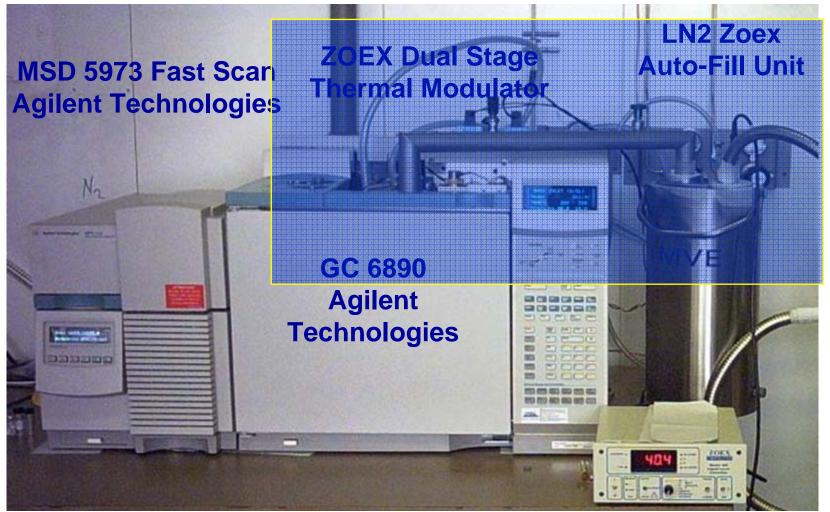
GC Agilent Technologies 6890N

- Injector Split/Splittless with EPC
- Split 1:100, carrier Helium
- Primary column: HP-5 MS (25m x 0.25mm ID, df=0.25μm)
- Secondary column: HP-WAX (0.7m x 0.1mm ID, df=0.1μm)
- Oven: 40°C x 0.5min, 1.5°C/min to 260°C, 260°C
 for 40 min (run time 187 min)
- Ramp Pressure: 20psi, 0.1psi/min to 25psi, 0.55psi/min to 55psi





GCxGC-MSD System







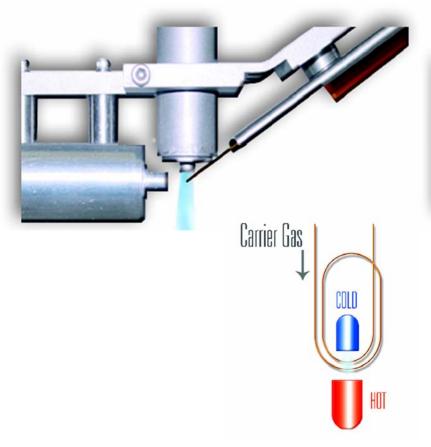
ZOEX dual stage thermal modulator

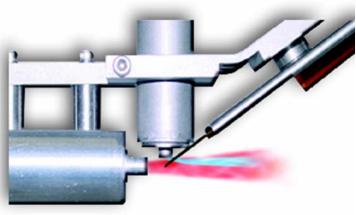
- KT-2004 Zoex Dual Stage Thermal Modulator
- LN2 Zoex Auto Fill Unit
- Zoex GC-Image Software
- Modulator Tube: uncoated fused silica tube 2mt X 0.1mmID
- Modulation period: 9 seconds
- Modulation duration: 450 msec
- Hot Jet Temperature: 350°C





ZOEX modulator operation



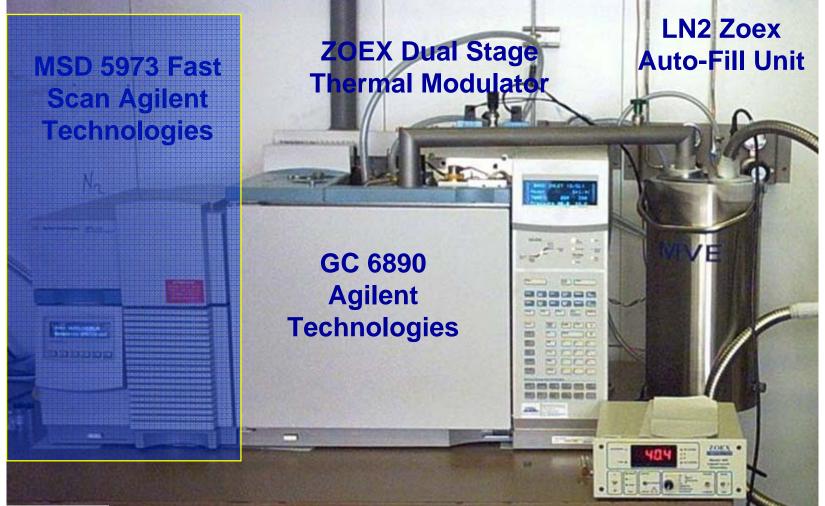








GCxGC-MSD System









MSD 5973 Fast Scan Agilent Technologies

- In order to maximize the spectral information, full-scan mass spectra acquisition mode (EI) was used
- MSD 5973 inert with Performance Electronics for Fast Scan Capabilities (10,000 amu/sec) and improved sensitivity
- Range 45-350 amu (19.80 scan/sec)
- MS Transfer line: uncoated fused silica tube (0.8m x 0.25mm ID, temperature 280°C)







Development of one GCxGC method for petrochemical analysis

How many variables in setting up a GCxGC method? Columns choice (stationary phase and lenght) Main oven temperature and/or secondary oven temperature Column flow Modulation period and duration Cold and Hot jet flows

<u>Simplify the user operation maintaining fixed the conditions</u> for all type of compounds in oil-derivatives







Column choice

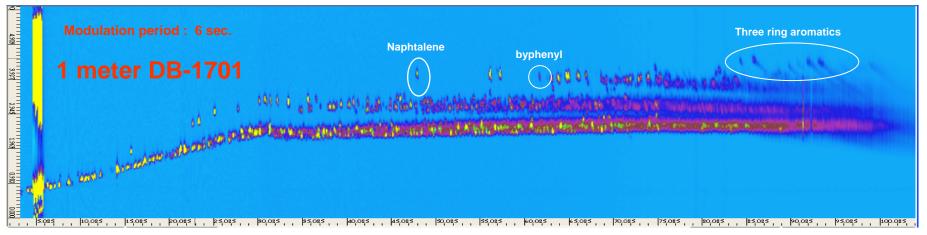
- Column combination to maximize GCxGC potentiality: the most "bi-dimensional" stationary phases must be used
 Primary column: apolar
 - Secondary column: polar
- Separate wide variety of hydrocarbon compounds (saturate/insaturate, light up to heavy aromatics and heteroatom compounds)
- Which stationary phase and column lenght?
- Wax : polyethylene glycol
- DB-1701: 14% cyanopropyl-phenyl, 86% dimethylpolysiloxane
- HP-17: 50% phenyl, 50% dimethylpolysiloxane



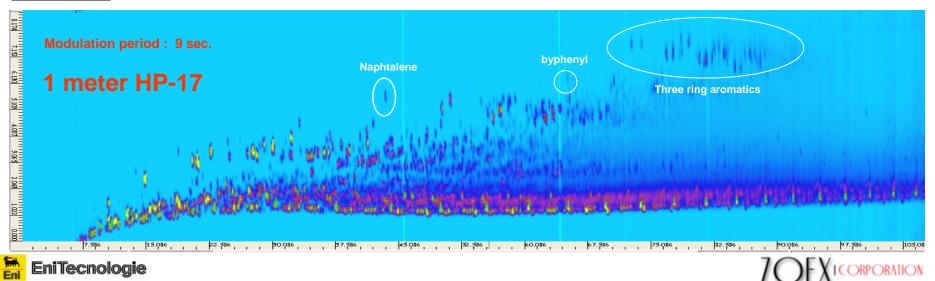




Secondary column 1 mt x 0.1 mmID x 0.1 um DB-1701Oven: 35°Cx0min, incr. 2°C/min to 300°C for 10 min No secondary oven

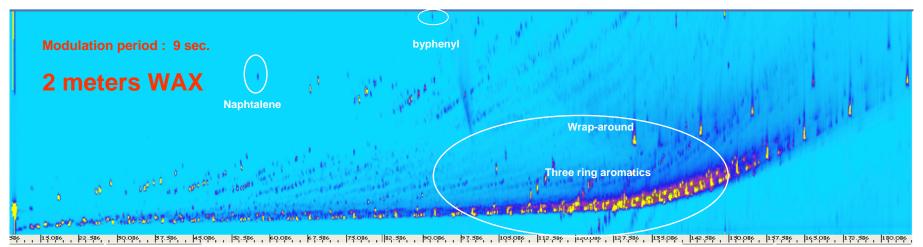


Secondary column <u>1 mt x 0.1 mmID x 0.1 um HP-17</u> Oven: 40°Cx0.5min, incr. 2°C/min to 300°C for 10 min <u>No secondary oven</u>

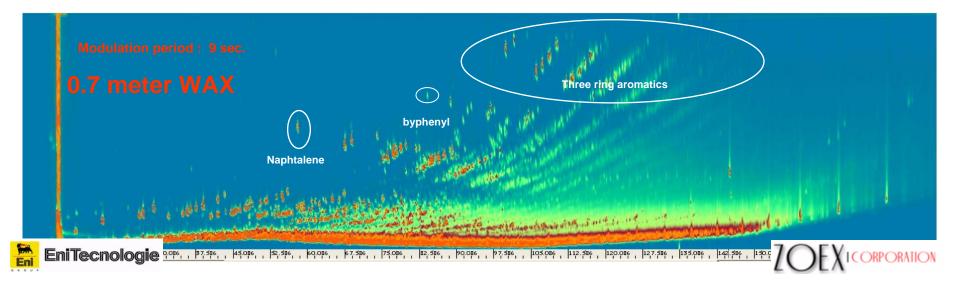




Secondary column <u>2 mt x 0.1 mmID x 0.1 um WAX</u> Oven: 40°Cx0.5min, incr. 0.5°C/min to 260°C for 40 min <u>Secondary Oven</u>: 30°C above main oven



Secondary column <u>1 mt x 0.1 mmID x 0.1 um WAX</u> Oven: 40°Cx0.5min, incr. 0.5°C/min to 260°C for 40 min <u>Secondary Oven</u>: 30°C above main oven



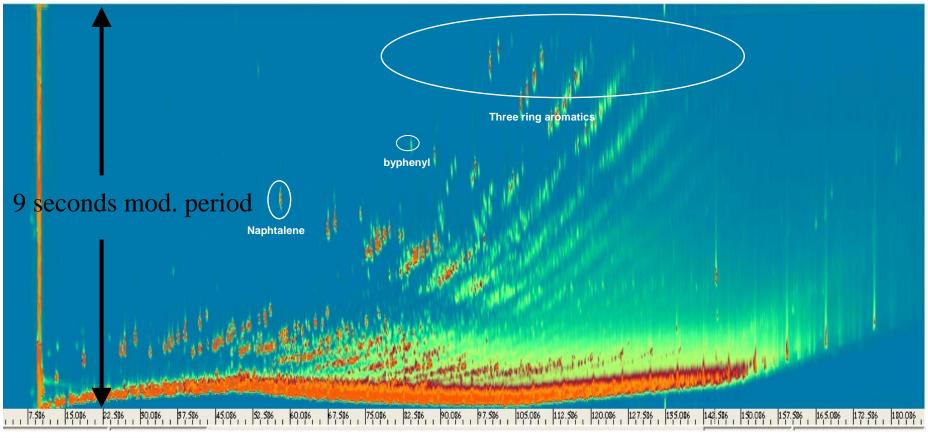


- Large enough to permit the elution of <u>all</u> the components from the secondary column (y- axis, separation by polarity)
- Should be tuned to collect at least three fraction of the eluting primary column peaks (high modulation period values require wide primary column peaks)





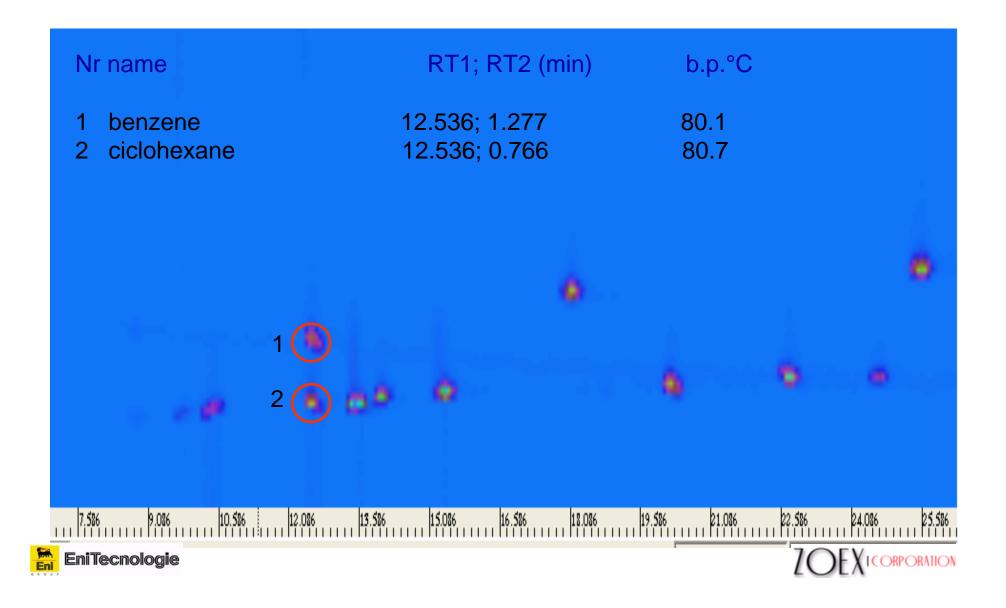




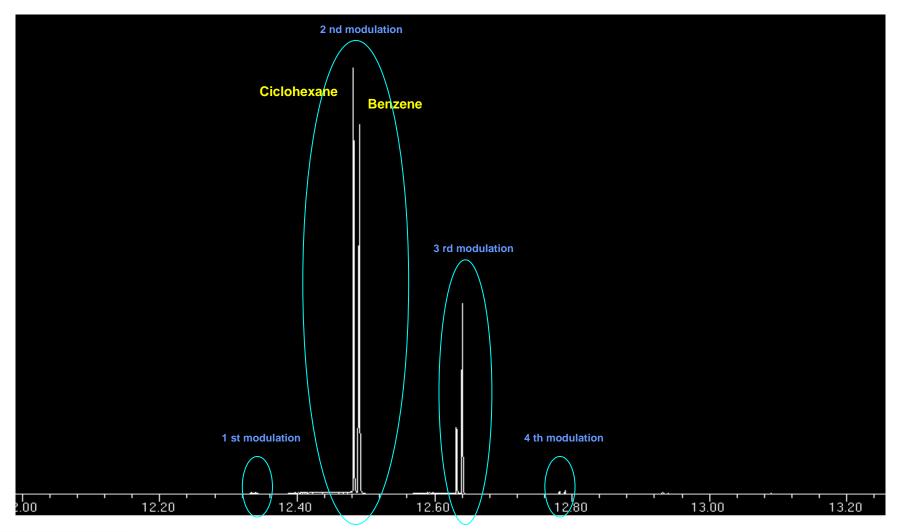


















Conclusions

 The GCxGC-qMSD showed to be a powerful tool in light petroleum fraction investigation. Its use allows the separation and identification of different polarity hydrocarbons (aromatic vs aliphatic) together with unique identification of heteroatom compounds, with a single sample run and without changing any analytical set up and parameters.









Thank you for your attention