

Carbonyl-DNPH Derivatives in Indoor and In-car Air by UHPLC and Triple Quadrupole LC/MS

Application Note

Environmental

Abstract

Ultra high performance liquid chromatography (UHPLC) and LC/MS/MS methods determine free carbonyl compounds in air after derivatization with 2,4-dinitrophenylhydrazine. The methods use UHPLC with DAD detection and triple quadrupole LC/MS/MS to identify and quantitate the target analytes. Using an Agilent Poroshell 120 EC-C18 HPLC column, UHPLC separates 14 carbonyl-DNPH components in 12 minutes. LC/MS/MS gives a lower limit of detection than LC alone. Using multiple reaction monitoring, more carbonyl-DNPH can be resolved.

Introduction

Carbonyl compounds are hazardous substances existing in air, with formaldehyde and acetaldehyde being the most abundant airborne carbonyls. They are classified as probable human carcinogens by the United States Environmental Protection Agency (EPA) and designated carcinogenic to humans by the International Agency for Cancer Research (IARC). The compounds are sources of contamination in indoor living and working environments, and are probably released from multiple sources, including plywood, particle board, furniture, glues, tobacco smoke, textiles, and other building materials. The air inside automobiles is another place where carbonyls are found, and many regulations from EU, Germany, China, and Japan are set to control levels of carbonyl compounds in vehicle air. Because of their highly volatile and reactive nature, carbonyl compounds are often converted to dinitrophenylhydrazine (DNPH) derivatives prior to LC analysis.



Authors

Rong-jie Fu, Maoxin Cao and Ying Wang Agilent Technologies (Shanghai) Co. Ltd. This application note developed a UHPLC method for separating 14 carbonyl-DNPH compounds in only 12 minutes. MS/MS detection was used to identify more compounds, because some could not be resolved through UHPLC alone. Both methods were developed on an Agilent Poroshell 120 EC-C18 column. Two samples collected from indoor air and in-vehicle air were analyzed, and some of carbonyl-DNPH compounds were detected and the amounts of them were calculated.

Materials and Methods

The Agilent 1290 Infinity LC System includes a binary pump, a thermostatted column compartment, a high-performance autosampler and a diode array detector. The LC/MS/MS system used a 1290 Infinity LC System and an Agilent 6460 Triple Quadrupole LC/MS.

Stock solutions were purchased from AccuStandard, Inc, New Haven, CT, USA (Table 1). They were diluted to appropriate concentrations with acetonitrile.

Two air samples were prepared as follows.

- Sample 6 L air using an AirChek 2000 air sampling pump, SKC Inc., Eighty Four, PA, USA, with DNPH-sampling SPE cartridge (silica coated with DNPH) at a flow rate of 200 mL/min in 30 minutes
- 2. Wash the cartridges with 5.0 mL acetonitrile
- Collect the eluted solution and filter through an Agilent 0.2 μm regenerated cellulose filter (p/n 5064-8221)
- 4. Transfer the filtered solutions to autosampler vials for analysis

Table 1 The target carbonyl-DNPH compounds analyzed in this application note.

No.	Name	CAS	Stock concentration (µg/mL)
1	Acetaldehyde-DNPH	1019-57-4	2.000
2	Acetone-DNPH	1567-89-1	2.000
3	Acrolein-DNPH	888-54-0	2.000
4	Benzaldehyde-DNPH	1157-84-2	2.000
5	2-Butanone-DNPH	958-60-1	2.000
6	Butyraldehyde-DNPH	1527-98-6	2.000
7	Crotonaldehyde-DNPH	1527-96-4	2.000
8	Cyclohexanone-DNPH	1589-62-4	5.000
9	Formaldehyde-DNPH	1081-15-8	4.000
10	Hexaldehyde-DNPH	1527-97-5	2.000
11	Methacrolein-DNPH	5077-73-6	2.000
12	Propionaldehyde-DNPH	725-00-8	2.000
13	p-Tolualdehyde-DNPH	2571-00-8	2.000
14	Valeraldehyde-DNPH	2057-84-3	2.000

Conditions

LC system

Column:	Agilent Poroshell 120 EC-C18, 3.0 × 75 mm, 2.7 μm (p/n 697975-302)			
Eluent:	A, Water; B, ACN			
Injection volume:	5 µL			
Flow rate:	1.0 mL/min			
Gradient:	Time (min) 0 2 7 12	% B 0 0 20 70		
Stop time:	14 minutes			
Post run:	1.5 minutes			
Temperature:	30 °C			
Instrument:	Agilent 1290 Infinity LC System (installed with a 1290 inline filter after injector valve, p/n 5067-4638)			
Detector:	UV, 360 nm			

LC/MS/MS system

Column:	Agilent Poroshell 120 EC-C18, 3.0 × 75 mm, 2.7 μm (p/n 697975-302)			
Eluent:	A, Water; B, ACN			
Injection volume:	5 µL			
Flow rate:	0.5 mL/min			
Gradient:	Time (min) 0 5 12 14	% B 35 55 70 70		
Temperature:	30 °C			
Stop time:	14 minutes			
Post run:	2.0 minutes			
Instrument:	Agilent 1290 LC Infinity (installed with a 1290 Infinity inline filter after injector valve, p/n 5067-4638), and an Agilent 6460 Triple Quadrupole LC/MS			
MS				
Gas temperature:	325 °C			
Gas flow:	10 L/min			
Nebulizer:	50 psi			
Sheath gas				

Nebulizer.	50 p3i
Sheath gas temperature:	350 °C
Sheath gas flow:	11 L/min
Capillary:	Negative 3,500 V

Optimized MRM conditions

Compound	lon source	lon pair qualitative and quantitative analyses		RT (min)
Formaldehyde	ESI (-)	208.8 → 163 (7 V)	208.8 → 133 (7 V)	4.084
Acetaldehyde	ESI (-)	223 → 163 (7 V)	223 → 122 (10 V)	5.203
Acetone	ESI (-)	237 → 178 (16 V)	237 → 164 (13 V)	6.329
Acrolein	ESI (-)	235 → 163 (10 V)	235 → 158 (10 V)	6.399
Propionaldehyde	ESI (-)	237 → 163 (10 V)	237 → 152 (12 V)	6.818
Crotonaldehyde	ESI (-)	249 → 202.1 (12 V)	249 → 172 (12 V)	7.710
Methacrolein	ESI (-)	249 → 172 (12 V)	249 → 109 (15 V)	8.079
2-Butanone	ESI (-)	251 → 178.2 (12 V)	251 → 164 (12 V)	8.137
Butyraldehyde	ESI (-)	251 → 163 (12 V)	251 → 122 (15 V)	8.364
Benzaldehyde	ESI (-)	285 → 238 (12 V)	285 → 163 (12 V)	9.052
Cyclohexanone	ESI (-)	277 → 247.2 (10 V)	277 → 231.1 (10 V)	9.710
Valeraldehyde	ESI (-)	265 → 191.1 (11 V)	265 → 163.1 (8 V)	10.087
<i>p</i> -Tolualdehyde	ESI (-)	299 → 252.1 (12 V)	299 → 163 (12 V)	10.729
Hexaldehyde	ESI (-)	279 → 163 (10 V)	279 → 205.2 (10 V)	11.917
	Formaldehyde Acetaldehyde Acetone Acrolein Propionaldehyde Crotonaldehyde Methacrolein 2-Butanone Butyraldehyde Benzaldehyde Cyclohexanone Valeraldehyde <i>p</i> -Tolualdehyde	FormaldehydeESI (-)AcetaldehydeESI (-)AcetoneESI (-)AcroleinESI (-)PropionaldehydeESI (-)CrotonaldehydeESI (-)CrotonaldehydeESI (-)ButyraldehydeESI (-)BenzaldehydeESI (-)CyclohexanoneESI (-)ValeraldehydeESI (-)ProjonaldehydeESI (-)EyraldehydeESI (-)EyraldehydeESI (-)EyraldehydeESI (-)EyraldehydeESI (-)EyraldehydeESI (-)P-TolualdehydeESI (-)	Formaldehyde ESI (-) 208.8 → 163 (7 V) Acetaldehyde ESI (-) 223 → 163 (7 V) Acetone ESI (-) 237 → 178 (16 V) Acrolein ESI (-) 235 → 163 (10 V) Propionaldehyde ESI (-) 237 → 163 (10 V) Crotonaldehyde ESI (-) 237 → 163 (10 V) Crotonaldehyde ESI (-) 249 → 202.1 (12 V) Methacrolein ESI (-) 249 → 172 (12 V) 2-Butanone ESI (-) 251 → 163 (12 V) Butyraldehyde ESI (-) 285 → 238 (12 V) Cyclohexanone ESI (-) 277 → 247.2 (10 V) Valeraldehyde ESI (-) 265 → 191.1 (11 V) <i>p</i> -Tolualdehyde ESI (-) 299 → 252.1 (12 V)	Formaldehyde ESI (-) $208.8 \rightarrow 163 (7 V)$ $208.8 \rightarrow 133 (7 V)$ Acetaldehyde ESI (-) $223 \rightarrow 163 (7 V)$ $223 \rightarrow 122 (10 V)$ Acetone ESI (-) $237 \rightarrow 178 (16 V)$ $237 \rightarrow 164 (13 V)$ Acrolein ESI (-) $237 \rightarrow 178 (16 V)$ $237 \rightarrow 164 (13 V)$ Acrolein ESI (-) $235 \rightarrow 163 (10 V)$ $237 \rightarrow 152 (12 V)$ Propionaldehyde ESI (-) $237 \rightarrow 163 (10 V)$ $237 \rightarrow 152 (12 V)$ Crotonaldehyde ESI (-) $249 \rightarrow 202.1 (12 V)$ $249 \rightarrow 172 (12 V)$ Methacrolein ESI (-) $249 \rightarrow 172 (12 V)$ $249 \rightarrow 109 (15 V)$ 2-Butanone ESI (-) $251 \rightarrow 178.2 (12 V)$ $251 \rightarrow 164 (12 V)$ Butyraldehyde ESI (-) $251 \rightarrow 163 (12 V)$ $251 \rightarrow 122 (15 V)$ Benzaldehyde ESI (-) $285 \rightarrow 238 (12 V)$ $285 \rightarrow 163 (12 V)$ Cyclohexanone ESI (-) $277 \rightarrow 247.2 (10 V)$ $277 \rightarrow 231.1 (10 V)$ Valeraldehyde ESI (-) $265 \rightarrow 191.1 (11 V)$ $265 \rightarrow 163.1 (8 V)$ p-Tolualdehyde ESI (-) $299 \rightarrow 252.1 (12 V)$ $299 \rightarrow 163 (12 V)$

Results and Discussion

The superficially porous particles of a Poroshell 120 have nearly identical efficiency as sub-2 µm totally porous materials and can be used to provide similarly fast and high resolution analyses at a lower pressure. A quick separation of 14 carbonyls in 12 minutes was achieved on the Poroshell 120 EC-C18 column with a gradient method (Figure 1), while a long analyses time of over 30 minutes was necessary to resolve all the compounds in another work [1].

Reasonable resolution was achieved between all the standard components, except methacrolein and 2-butanone. These two compounds could still be identified because of different product ions using the triple quadruple LC/MS system. Reproducibility tests were carried out by seven consecutive injections of 0.5 ng standards. This separation could be easily reproduced even at such low carbonyl levels.







- Formaldehyde
- Acetaldehyde
- Acetone
- Acrolein
- Propionaldehyde
- Crotonaldehyde
- Methacrolein
- 2-Butanone
- Butyraldehyde
- 10. Benzaldehyde
- 11. Cyclohexanone
- 12. Valeraldehyde
- 13. p-Tolualdehyde
- 14. Hexaldehyde



Figure 2. Overlay chromatogram of seven consecutive injections of 0.5 ng standards on an Agilent Poroshell 120 EC-C18 column.

The air samples from indoor and in-vehicle air were measured by the UHPLC method. Figure 3 shows chromatograms of sample 1, sample 2, and the standards. Compounds varied in the different samples. Acetaldehyde, acetone, and acrolein were found in the two samples but additional compounds were detected in sample 2. The amounts of the detected compounds could be measured given the concentration of the standards.



Figure 3. Chromatograms of carbonyl samples and standards using an Agilent Poroshell 120 EC-C18 column.

The LC method was then transferred to an LC/MS/MS method because the mobile phase using water/acetonitrile was MS friendly. Compounds such as methacrolein and 2-butanone that could not be resolved by LC could be identified by LC/MS/MS due to the different product ions. Figure 4 shows the MRM (multiple-reaction monitoring) chromatograms of 14 carbonyl-DNPH standards and a sample of in-vehicle air. The calculated limit of detection with triple quadrupole LC/MS was in the range of 0.03 to 0.3 parts per billion (ppb), which was far below that of the UHPLC method of 60 to 180 ppb.



Figure 4. MRM chromatograms of carbonyl standards and sample (in-vehicle air) using an Agilent Poroshell 120 EC-C18 column.

Conclusions

A method was developed for the separation of carbonyl components using the Agilent Poroshell 120 EC-C18 column. The column provided good selectivity for these compounds and good resolution. The method developed on the Agilent 1290 Infinity LC was suitable for fast screening and quantitative analysis of these compounds in air. The LC/MS/MS method enabled co-eluted compounds to be identified and quantified using MRM and was suitable for low-level concentration analysis of carbonyls.

Reference

1. Anon. Determination of Carbonyl Compounds by High Performance Liquid Chromatography (HPLC), EPA Method 8315a. Environmental Protection Agency, Washington, DC, USA (1996).

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